E) Sample Chemical "Functional Groups" Associated With Chemicals Addressed in This Matter.

#### **Exhibit E:**

### **List of Chemical Functional Groups Common In Contraband**

### 1) Hydrocarbons

Chemical class	Group	Formula	Structural Formula	Prefix	Suffix	Example
Alkane	Alkyl	RH	$\mathbb{R}^{\left\langle \cdot \right\rangle_{n}}$	alkyl-	-ane	H H H Ethane
Alkene	Alkenyl	$R_2C=CR_2$	$R_1$ $R_3$ $R_2$ $R_4$	alkenyl-	-ene	H H C=C' H' H  Ethylene (Ethene)
Alkyne	Alkynyl	RC≡CR'	R———R'	alkynyl-	-yne	H—C≡C—H Acetylene (Ethyne)
Benzene derivative	Phenyl	$rac{ ext{RC}_6 ext{H}_5}{ ext{RPh}}$	R—	phenyl-	-benzene	Cumene (2-phenylpropane)

In all the examples shown, the designation "R" means "an organic substituent". It may be the same or some other functional group, or may represent the remainder of a given molecule relative to a spercific functional group.

# 2) Groups Containing Oxygen

Chemical class	Group	Formula	Structural Formula	Prefix	Suffix	Example
Alcohol	Hydroxyl	ROH	R—Q H	hydroxy-	-ol	H H Methanol
Ketone	Carbonyl	RCOR'	R R'	-oyl- (- COR') or oxo- (=O)	-one	Butanone (Methyl ethyl ketone)
Aldehyde	Aldehyde	RCHO	R H	formyl- (- COH) or oxo- (=O)	-al	H Ethanal (Acetaldehyde)
Carbonate	Carbonate ester	ROCOOR	R <sub>1</sub> O R <sub>2</sub>	(alkoxycarb onyl)oxy-	alkyl carbonate	Triphosgene (Di(trichloromethyl) carbonate)
Carboxylate	Carboxylate	RCOO⁻	R O	carboxy-	-oate	Sodium acetate (Sodium ethanoate)
Carboxylic acid	Carboxyl	RCOOH	ROH	carboxy-	-oic <b>acid</b>	OH Acetic acid (Ethanoic acid)
Ester	Ester	RCOOR'	R OR'	alkanoylox y- or alkoxycarb onyl	alkyl alkan <b>oate</b>	Ethyl butyrate (Ethyl butanoate)
Peroxide	Peroxy	ROOR	R 0-0 R'	peroxy-	alkyl <b>peroxide</b>	Di-tert-butyl peroxide
Ether	Ether	ROR'	R <sup>O</sup> R'	alkoxy-	alkyl <b>ether</b>	Diethyl ether (Ethoxyethane)

# 3) Groups Containing Nitrogen

Chemical class	Group	Formula	Structural Formula	Prefix	Suffix	Example
Amide	Carboxamide	RCONR <sub>2</sub>	R"R"	carboxamido- or carbamoyl-	-amide	O NH <sub>2</sub> Acetamide (Ethanamide)
Amines	Primary amine	$\mathbf{RNH}_2$	R <sup>N</sup> H H	amino-	-amine	H H H H Methylamine (Methanamine)
	Secondary amine	$R_2NH$	R-N R'	amino-	-amine	$H$ $CH_3$ $CH_3$ Dimethylamine
	Tertiary amine	$\mathbf{R}_3\mathbf{N}$	R-NR'	amino-	-amine	N       Trimethylamine
	4° ammonium ion	$R_4N^+$	R <sub>1</sub> /N <sup>+</sup> R <sub>1</sub> /N <sup>+</sup> R <sub>2</sub>	ammonio-	-ammonium	[ →N <sup>+</sup> →OH ] X-
Nitrate	Nitrate	RONO <sub>2</sub>	R_0_N^+_0_	nitrooxy-, nitroxy-	alkyl <b>nitrate</b>	Amyl nitrate (1-nitrooxypentane)
Nitrite	Nitrosooxy	RONO	R_0/N_0	nitrosooxy-	alkyl <b>nitrite</b>	Isoamyl nitrite (3-methyl-1- nitrosooxybutane)
Pyridine derivative	Pyridyl	RC₅H₄N	$\begin{array}{c c} R & & N \\ \hline R & & N \\ \hline R & & N \\ \hline \end{array}$	4-pyridyl (pyridin-4-yl)  3-pyridyl (pyridin-3-yl)  2-pyridyl (pyridin-2-yl)	-pyridine	Nicotine